

Solvents

WHAT ARE SOLVENTS?

Solvents are any class of chemicals used to dissolve materials. Usually, solvents are extremely volatile and reactive, meaning they quickly evaporate into their gaseous form and interact with other molecules to form byproducts (or different chemical combinations) quickly.¹

Because of their ability to dissolve solids, solvents are often used to clean and sanitize. Solvents are also used industrially for many purposes such as degreasing equipment or cleaning electronics before sale.¹

HOW ARE WE EXPOSED?

People can be exposed to solvents in a number of ways:

- The main route of exposure for solvents is breathing it in.
- Household exposures to solvents are fairly common. Solvents are commonly added to many household products such as cleaning supplies and paints.¹
- Individuals may also be exposed to solvents in the workplace, while working with solvents.¹
- Indirect exposure can occur when solvents added to consumer products—such as paint or gasoline—evaporate into the air.
- Environmental exposures to solvents are also possible, as some solvents persist in the environment due to improper disposal.² Other solvents such as benzene are released into the air from gasoline exhaust.³

SYMPTOMS & HEALTH OUTCOMES

There are different health effects of solvent exposure depending on the specific solvent and the length of exposure. Different solvents will result in a range of symptoms. The health effects of some common solvents are included here.

Ammonia

Ammonia is commonly occurring and naturally present in the environment. Exposure to these background levels is not harmful.⁴ However, important research shows that acute exposure to high concentrations of ammonia and chronic exposure to moderate levels can be harmful or deadly.

At low levels, ammonia inflicts tissue damage to the exposed area, including the nose, throat, and lungs if it is inhaled.^{5,6} At higher levels, animal studies suggest it may interfere with neurological processes, especially when ingested.^{7,8} Airborne ammonia can cause irritation of the eyes resulting in inflammation, short-term blindness or altered vision, watering eyes, and corneal abrasions.⁴

Bleach

Sodium hypochlorite is the active ingredient in chlorine bleach and cleaning products containing bleach. Exposure to sodium hypochlorite irritates skin and mucus membranes, and high concentrations of sodium hypochlorite can cause serious tissue damage.⁹

In addition to being acutely toxic, sodium hypochlorite can react with other cleaners and organic matter to form dangerous chemicals. Chloroform can be created when bleach contacts organic matter during cleaning¹⁰ and can be created in chlorinated drinking water as well.¹¹ Free chlorine gas can be created when bleach is mixed with acidic solutions.¹²

Chloroform is a probable human carcinogen according to the US EPA.¹³ Chronic exposure is known to cause depression, lack of concentration, and irritability. Chlorine gas is acutely toxic and can cause irritation to the nose and throat, chest pain, vomiting, and even death.¹⁴

Benzene

Benzene is a petrochemical, meaning it is a refined product of oil. It is present in the environment in water, soil and air. Some of the major sources of benzene are car exhaust, cigarette smoke, and paint or glues.³ Exposure to benzene through food, water, and soil is not common, unless there is a local source of contamination. Approximately 95% of exposure to benzene is through inhalation.¹⁵ Acute exposure can cause dizziness, rapid heart rate, tremors and unconsciousness.³

Benzene is known to cause cancer. The most significant health risk to low level exposure to benzene is leukemia.¹⁵ However, research suggests that benzene can induce other diseases of the blood, such as anemia, and uncontrolled bleeding.¹⁶ Benzene may also cause reproductive effects such as irregular menstruation and birth defects.³ (*Please see the Benzene fact sheet for more details.*)

Simple Green

Simple Green contains 2-butoxyethanol, a type of chemicals known as ethylene glycol

ethers (EGEs), that are commonly used as solvents both in businesses and industries.

In humans, long term exposure to EGEs can result in dizziness, lethargy, headaches, and memory loss as well as hematopoietic effects such as decreases in some types of blood cells.^{17,18} Long-term exposure in animals has been linked to damage to DNA, sperm and blood cells.^{19,20,21} Exposure to 2-butoxyethanol is known to cause hemolysis, or destruction of blood cells, and reduce concentrations of hemoglobin in rats,²² and research has correlated 2-butoxyethanol exposure to an increase in tumors in rats and mice.²³

Toluene

Exposure to toluene can happen in many different ways, but the main route of exposure is inhalation from contaminated air. Gasoline, car exhaust, cigarette smoke, and paint or paint thinner are all common sources of toluene.²⁴ Using nail polish significantly increases concentrations of toluene in the air.²⁵ Acute exposure to toluene has been shown to result in dizziness, headaches,²⁶ manual impairment, intoxication, and loss of color recognition.²⁷

In humans, long-term exposure to toluene has been linked to mucus membrane irritation, altered temperature and sound perception, intoxication²⁴ and an increase in spontaneous abortions.²⁸ It is also linked to dizziness, headaches,²⁹ and hearing loss.³⁰ In animals, toluene exposure has been linked to lower dopamine and noradrenaline levels, important neural hormones³¹ and an overall decrease in brain weight.³²

FOLLOW UP ACTION

- If you think your patient may have been exposed to solvents, refer patient to a physician.

REDUCING OUR EXPOSURE

You can prevent or minimize exposure to solvents in the following ways:

- Whenever possible, buy products that do not contain harmful solvents. Water-based paints and glues and citrus-based cleaners are safer to use.
- If you have solvents or products containing solvents in your home, carefully label them and store them in a safe place, especially in homes with young children.
- When you use products with solvents, such as paints, cleaners, or nail polish, make sure they are used in well ventilated areas and open windows or use fans to reduce the amount of solvents in the air.
- Never mix cleaning solutions or solvents, as the resulting vapors can be harmful or deadly.
- Gasoline and exhaust are major sources of solvents and other contaminants. Turn off the engines of cars, ATVs, and snow machines when not needed.

¹ National Institute for Occupational Safety and Health (NIOSH). 2009. NIOSH Safety and Health Topic: Organic Solvents. Available: <http://www.cdc.gov/niosh/topics/organsolv/>.

² Yeh HC, Kastenber W. 1991. Health risk assessment of biodegradable volatile organic chemicals: A case study of PCE, TCE, DCE and VC. *J. Hazard. Mater* 27:111–126.

³ Agency for Toxic Substances and Disease Registry (ATSDR). 2007. Toxic substance profiles: Benzene. Available: <http://www.atsdr.cdc.gov/toxprofiles/phs3.html>.

⁴ Agency for Toxic Substances and Disease Registry (ATSDR). 2004. Toxic substance profiles: Ammonia. <http://www.atsdr.cdc.gov/toxprofiles/tp126.html>.

⁵ MacEwen JD, Theodore J, Vernot EH. 1970. Human exposure to EEL concentrations of monomethylhydrazine. In: *Proceedings 1st Annual Conference on Environmental Toxicology*, pp. 355-363. Wright-Patterson Air Force Base, OH: Aerospace Medical Research Laboratory.

⁶ Verberk MM. 1977. Effects of ammonia in volunteers. *Int Arch Occup Environ Health* 39:73-81.

⁷ Tepper JS, Weiss B, Wood RW. 1985. Alterations in behavior produced by inhaled ozone or ammonia. *Fundam Appl Toxicol.* 5:1110-1118.

⁸ Coon RA, Jones RA, Jenkins LJ Jr. 1970. Animal inhalation studies on ammonia, ethylene, glycol, formaldehyde, dimethylamine and ethanol. *Toxicol Appl Pharmacol* 16:646-655.

⁹ Agency for Toxic Substances and Disease Registry (ATSDR). 1997. Toxic substance profiles: Sodium hypochlorite. Available: <http://www.atsdr.cdc.gov/tfacts184.html>.

¹⁰ Odabasi M. 2008. Halogenated Volatile Organic Compounds from the Use of Chlorine-Bleach-Containing Household Products. *Environ. Sci. Technol.* 42(5):445–1451.

¹¹ Kavcar P, Odabasi M, Kitis M, Inal F, Sofuoglu SC. 2006. Occurrence, oral exposure and risk assessment of volatile organic compounds in drinking water for Izmir. *Water Res.* 40:3219-3230

¹² Nazaroff WW, Weschler CJ. 2004. Cleaning products and air fresheners: Exposure to primary and secondary air pollutants *Atmos. Environ.* 38:2841-2865.

¹³ United States Environmental Protection Agency (EPA). 2008. Integrated Risk Information System: Chloroform. Available: <http://www.epa.gov/iris/subst/0025.htm>.

¹⁴ Agency for Toxic Substances and Disease Registry (ATSDR). 2007. Toxic substance profiles: Chlorine. Available: <http://www.atsdr.cdc.gov/toxprofiles/tp172.html>.

¹⁵ Duarte-Davidson R, Courage C, Rushton L, Levy L. 2001. Benzene in the environment: An assessment of the potential risks to the health of the population. *Occup Environ Med* 58:2-13.

¹⁶ Smith MT. 1996. The mechanism of benzene-induced leukemia: A hypothesis and speculations on the causes of leukemia, *Environ Health Perspect* 104(6):1219–1225.

¹⁷ Browning RG, Curry SC. 1994. Clinical toxicology of ethylene glycol ethers. *Hum. Exp. Toxicol.* 13:325–335.

¹⁸ Kalf GF, Post GB, Snyder R. 1987. Solvent toxicology: Recent advances in the toxicology of benzene, the glycol ethers, and carbon tetrachloride. *Annu. Rev. Pharmacol. Toxicol.* 27:399–427.

¹⁹ Johanson G. 1988. Aspects of biological monitoring of exposure to glycol ethers. *Toxicol. Lett.* 43:5–21.

²⁰ Johanson G. 2000. Toxicity review of ethylene glycol monomethyl ether and its acetate ester. *Crit. Rev. Toxicol.* 30:307–345.

-
- ²¹ Ghanayem BI, Burka LT, Matthews HB. 1987. Metabolic basis of ethylene glycol monobutyl ether (2-butoxyethanol) toxicity: Role of alcohol and aldehyde dehydrogenases. *J. Pharmacol. Exp. Ther.* 242:222-231.
- ²² Ghanayem BI, Ward SM, Chanas B, Nyska A. 2000. Comparison of the acute hematotoxicity of 2-butoxyethanol in male and female F344 rats. *Human & Experimental Toxicology.* 19:85-192.
- ²³ National Toxicology Program. 2000, Mar. Toxicology and carcinogenesis studies of 2-Butoxyethanol (CAS No. 111-76-2) in F344/N rats and B6C3F1 mice (inhalation studies). *National Toxicology Program Technical Report Series* 484:1-290. Bethesda, MD: Department of Health and Human Services, Public Health Service.
- ²⁴ Agency for Toxic Substances and Disease Registry (ATSDR). 2000. Toxicological profiles: Toluene. Available: <http://www.atsdr.cdc.gov/toxprofiles/phs56.html>.
- ²⁵ Curry KK, Brookman DJ, Whitmyre GK. 1994. Personal exposures to toluene during use of nail lacquers in residences: Description of the results of a preliminary study. *J Expos Anal Environ Epidemiol* 4(4):443-456.
- ²⁶ Anderson MA. 1992. Influence of surfactants on vapor-liquid partitioning. *Environ Sci Technol* 26:2186-2191.
- ²⁷ Baelum J, Andersen I, Lundqvist GR. 1985. Response of solvent-exposed printers and unexposed controls to six-hour toluene exposure. *Scand J Work Environ Health* 11:271-280.
- ²⁸ Ng TP, Foo SC, Yoong T. 1992. Menstrual function in workers exposed to toluene. *Br J Ind Med* 49:799-803.
- ²⁹ Yin S, Li G, Hu Y, et al. 1987. Symptoms and signs of workers exposed to benzene, toluene or the combination. *Ind Health* 25:113-130.
- ³⁰ Morata TC, Fiorini AC, Fischer FM. 1997. Toluene-induced hearing loss among rotogravure printing workers. *Scand J Work Environ Health.* 23(4):289-98.
- ³¹ Ikeda M, Koizumi A, Kasahara M. 1986. Combined effects of n-hexane and toluene on norepinephrine and dopamine levels in rat brain tissues after long-term exposure. *Bull Environ Contam Toxicol.* 36:510-517.
- ³² Kyrklund T, Kjellstrand P, Haglid K. 1987. Brain lipid changes in rats exposed to xylene and toluene. *Toxicology* 5:123-133.